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1,3-Polyol arrays via the stereoselective rearrangement of vinyl acetals

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Abstract—A series of 1,3-dioxanyl vinyl acetals were readily synthesized from the corresponding dioxanone by a reduction and in situ acylation followed by Petasis olefination. Treatment of these vinyl acetals with $BF_3 \cdot OEt_2$ results in an O to C rearrangement to form *anti*-3,5-dihydroxyketones while a mixture of Me_3Al and $BF_3 \cdot OEt_2$ provides the corresponding *syn* relationship via a stereoretentive rearrangement. © 2003 Elsevier Ltd. All rights reserved.

1. Introduction

Skipped polyol units are ubiquitous structural motifs in several important classes of natural products. For example, the polyene macrolide class of natural products is characterized by a conjugated polyene and skipped polyol unit.¹ Due to their important biological activity and structural complexity, numerous efforts have been directed at the development of an efficient synthetic method for stereoselective synthesis of 1,3-polyol segments.²

Recently, we have developed a Lewis acid-mediated highly stereoretentive rearrangement of vinyl acetals, wherein the selectivity is controlled by tight ion pairing in the solvent cage.³ In many cases, either isomer can be prepared by simple choice of Lewis acid systems (BF₃·OEt₂ or Me₃Al/ $BF_3 \cdot OEt_2$). In order to explore this unusual methodology fully, we have been interested in new methods for the stereoselective synthesis of skipped polyols employing the Lewis acid-promoted rearrangement of vinyl acetals. Herein we report that the Lewis acid-mediated rearrangement of dioxanyl-derived vinyl acetals is an effective method for the highly stereocontrolled synthesis of both syn- and anti-3,5dihydroxy ketone units (Eq. (1)). The presence of two distinct acetal functionalities in these substrates was a potential cause for concern, but we expected to disfavor ring acetal binding by increasing the bulk of the R group. A subsequent diastereoselective addition to the newly formed ketone would provide a stereoselective entry to 1,3,5-triol fragments.



2. Results and discussion

2.1. Substrate synthesis

1,3-Dioxanones were efficiently prepared from β -hydroxy acids utilizing the protocol described by Crich et al.⁴ Thus, reaction of a β -hydroxy acid and suitable aldehyde, isopropoxytrimethylsilane, and TMSOTf in the presence of 4 Å molecular sieves at -78° C to -15° C for 8 h afforded the corresponding 1,3-dioxanone in moderate to good yield after purification by silica gel chromatography (Scheme 1). The dioxanones are typically formed with better than 10:1 selectivity at the acetal carbon.

A one-pot reduction/acylation protocol developed by Rychnovsky⁵ provides a general synthesis to the 4-acyloxy-1,3-dioxane required for this study. Yields for this reaction typically exceed 80% and the products are easily purified by flash silica gel chromatography. It should be noted that only trace product was formed if benzoyl chloride was used as an acylating agent. Finally, the acylation products were converted to the corresponding vinyl acetals in moderate to good yield by the action of Cp_2TiMe_2/Cp_2TiCl_2 at 80°C for 12 h.⁶

2.2. Stereoselective rearrangement of vinyl acetals

Initial attempts to effect the rearrangement of a vinyl acetal

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Scheme 1.

Table 1. Effect of 2-substitutent on rearrangement

$\begin{array}{c} & & \\$	Lewis Acid PhMe –78 °C, 12 h	$\frac{1}{R^{1}}$	+ O		(2)
Entry	Vinyl acetal		Product	syn/anti (isolated yield) ^a	
				BF ₃ ·OEt ₂ ^b	Me ₃ Al/BF ₃ ·OEt ₂ ^c
1 2 3 4 5	5a (R^1 =Me, R^2 =H) 5b (R^1 =Ph, R^2 =H) 5c (R^1 =Me, R^2 =Me) 5d (R^1 = <i>i</i> -Pr, R^2 =H) 4aa (R^1 = <i>i</i> -Bu, R^2 =H)		6a 6b 6c 6d 7aa	Many products Many products 16:84 (65%) 3:97 (92%) 2:98 (82%)	Many products Many products Many products 90:10 (80%) 96:4 (90%)

^a Selectivity determined by ¹H NMR analysis of the unpurified reaction mixture; isolated yield indicates analytically pure material of the major isomer.

^b BF₃·OEt₂ (1.2 equiv.) used as Lewis acid.

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^c Me₃Al (4.0 equiv.) and BF₃·OEt₂ (1.2 equiv.) used as Lewis acid.

with a methyl group at the 2-position revealed that the $Me_3Al/BF_3 \cdot OEt_2$ -mediated rearrangement produced many products. In the case of $BF_3 \cdot OEt_2$, epimerization occurred at the dioxane ring acetal, complicating product analysis. These results are summarized in Table 1. Although isobutyraldehyde-derived dioxanyl acetal **5d** provided reasonable levels of selectivity in the rearrangement

reaction (entry 4, Table 1), the larger *tert*-butyl group provided enough of a benefit to warrant its use. The use of $BF_3 \cdot OEt_2$ as a Lewis acid provides the *anti* relationship selectively, while Me₃Al/BF₃·OEt₂ provides *syn*. Relative stereochemistry was determined by nOe experiments. Having identified the *tert*-butyl group at the 2-position as providing optimal selectivities, the scope and limitations of

Table 2. Scope of	the vinyl acetal rearrangement—dios	anyi tragment		
	PhMe PhMe Ph	+ O O O O		(3)
4aa-4fa	syn- 7aa-7fa	anti- 7aa-7fa		
Entry ^a	Vinyl acetal	Product	syn/anti (isolated yield) ^b	
			$BF_3 \cdot OEt_2^c$	Me ₃ Al/BF ₃ ·OEt ₂ ^d
1	4aa (R=Me)	7aa (R=Me)	2:98 (82%)	96:4 (90%)
2	4ba (R= <i>n</i> -Hex)	7ba (R= <i>n</i> -Hex)	2:98 (87%)	92:8 (90%)
3	4ca (R=cy-Hex)	7ca ($R=cy$ -Hex)	2:98 (96%)	93:7 (87%)
4	4da $(R=i-Pr)$	7da (R= <i>i</i> -Pr)	2:98 (97%)	93:7 (85%)
5 4ea (R=Ph)		7ea (R=Ph)	2:98 (87%)	92:8 (81%) ^e
$6 \qquad 4 fa (R = CH_2 CH_2 Ph)$		7fa (R=CH ₂ CH ₂ Ph)	2:98 (91%)	98:2 (89%) ^f

^a All reactions conducted in toluene at -78°C for 12 h on 0.1 mmol scale, unless otherwise stated.

^b Selectivity determined by ¹H NMR analysis of the unpurified reaction mixture; isolated yield indicates analytically pure material of the major isomer.

^c BF₃·OEt₂(1.2 equiv.) used as Lewis acid.

^d Me₃Al (4.0 equiv.) and BF₃·OEt₂ (1.2 equiv.) used as Lewis acid.

Reaction allowed to stir for 36 h.

^f Reaction carried out on 1 mmol scale.

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Table 3. Scope of the vinyl acetal rearrangement—enolate fragment $Ph \longrightarrow 0 \longrightarrow R$ Lewis Acid $Ph \longrightarrow 0 \longrightarrow R$ Lewis Acid $Ph \longrightarrow 0 \longrightarrow R$ $Ph \longrightarrow 0 \longrightarrow R$ $Ph \longrightarrow 0 \longrightarrow 0$ $Ph \longrightarrow 0$ $Ph \longrightarrow 0 \longrightarrow 0$ $Ph \longrightarrow 0$						
	4fb-4fc	syn- 7fb-7fc	anti- 7fb-7fc			
Entry ^a	Vinyl acetal	Product	Temp (°C)	syn/anti (isolated yield) ^b		
				BF ₃ ·OEt ₂ ^c	Me ₃ Al/BF ₃ ·OEt ₂ ^d	
1	4fb (R=Me)	7fb (R=Me)	-78	17:83 (78%)	67:33 (ND)	
2	4fb (R=Me)	7fb (R=Me)	-55	_	96:4 (89%)	
3	4fc (R=Et)	7fc (R=Et)	-78	4:96 (91%)	78:22 (ND)	
4	4fc (R=Et)	7fc (R=Et)	-55	-	92:8 (88%)	

^a All reactions conducted in toluene at the indicated temperature for 12 h, unless otherwise stated.

^b Selectivity determined by ¹H NMR analysis of the unpurified reaction mixture; isolated yield indicates analytically pure material of the major isomer.

^c BF₃·OEt₂ (1.2 equiv.) used as Lewis acid.

^d Me₃Al (4.0 equiv.) and BF₃·OEt₂ (1.2 equiv.) used as Lewis acid.

the rearrangement reaction were further studied with vinyl acetals containing this substituent (Table 2).

A variety of sterically and electronically dissimilar substituents at the 6-position could undergo rearrangement with high selectivity and good yield in the presence of Me₃Al/BF₃·OEt₂ or BF₃·OEt₂. When Me₃Al/BF₃·OEt₂ was used, the rearrangement reaction leads to formation of syn-1,3 product. The use of $BF_3 \cdot OEt_2$ provides the corresponding anti adduct. In the case of vinyl acetal 4fb, warming the reaction temperature led to greatly improved selectivity, affording the syn-product in 96:4 selectivity (Table 3). This trend was further reinforced by vinyl acetal 4fc, which rearranged at higher temperature with much better selectivity. This is consistent with our rationale of the stereochemistry of Me₃Al/BF₃·OEt₂-mediated rearrangement of pyranyl vinyl acetals, where the selectivity could be controlled by tight ion pairing.³ At higher temperature, tight ion pairing is favored relative to solvent-dissociated ions for entropic reasons.⁷

To further demonstrate the synthetic utility of our methodology, we investigated the rearrangement of 5-methyl vinyl acetals **11** and **15**. Mixed polypropionatepolyacetate arrays are common units in biologically important compounds.⁸ Many elegant approaches have been developed, yet the construction of these units remains a challenge.⁹ We envisioned that the Lewis-acid-mediated rearrangement of an appropriately substituted vinyl acetal could provide a new entry. To this end, we prepared the rearrangement precursors **11** and **15** employing routine synthetic sequences (Scheme 2).

The diastereomeric vinyl acetals **11** and **15** were each subjected to the two sets of reaction conditions described above (Scheme 3). Substrate **11** bearing the 1,2-*anti* relationship underwent highly selective rearrangement, affording 1,3-*anti*-**17** in 98:2 selectivity using $BF_3 \cdot Et_2O$ and 1,3-*syn*-**16** in 96:4 selectivity using $Me_3Al/BF_3 \cdot OEt_2$. Rearrangement of substrate **15** was equally effective in the presence of $BF_3 \cdot OEt_2$. The 1,3-*anti* isomer **19** was formed in 98:2 selectivity. However, treatment of this acetal with $Me_3Al/BF_3 \cdot OEt_2$ resulted in no reaction, even at ambient temperature.

We believe that the lack of reactivity between substrate 15



a) pivalaldehyde, i-PrOTMS, 4Å MS, TMSOTf, CH₂Cl₂; b) DIBALH; (PhCO)₂O, Py, DMAP; c) Cp₂TiMe₂, Cp₂TiCl₂.



Scheme 3.





and Me₃Al/BF₃·OEt₂ may provide some insight as to the mechanism of ionization. Vinyl acetals **11** and **15** are presumably in chair conformations with the large *tert*-butyl group equatorial (Scheme 4). Coordination of Lewis acid to the enol ether oxygen forms **20** and **21**, respectively. The latter suffers from *syn*-pentane interactions with the axial methyl group, potentially disfavoring its formation altogether. Once the Lewis acid is coordinated to the enol ether oxygen, ionization of the enolate presumably has to occur at an axial position. We suggest that torsional strain with the *cis*-5-methyl group prevents this from occurring.

In conclusion, we have developed a new process for highly stereocontrolled synthesis of both *syn-* and *anti-3*,5-di-hydroxy ketone units via O to C rearrangement of vinyl acetals. The reactions described here significantly expand the scope of stereoretentive O to C rearrangement of vinyl

acetals. Further investigations of this reaction and applications to the efficient synthesis of stereochemically complex natural products are ongoing.

3. Experimental

3.1. General

All reactions were carried out under an atmosphere of argon in flame-dried glassware with magnetic stirring. Tetrahydrofuran, diethyl ether, and dichloromethane were degassed with argon and passed through two columns of neutral alumina. Toluene was degassed with argon and passed through one column of neutral alumina and one column of Q5 reactant. Column chromatography was performed on EM Science silica gel 60 (230–400 mesh). Thin layer chromatography was performed on EM Science 0.25 mm silica gel 60-F plates. Visualization was accomplished with UV light, KMnO₄, aqueous ceric ammonium molybdate, or bromocresol green dips followed by heating.

Melting points were measured with a MelTemp II melting point apparatus outfitted with a Fluke 51 thermocouple and are uncorrected. Infrared spectra were obtained on a Nicolet Avatar 320 FT-IR spectrometer. ¹H NMR and spectra were recorded on a Varian 300, 400, or 500 MHz spectrometer at ambient temperature. Data are reported as follows: chemical shift in parts per million (δ , ppm) from an internal standard [tetramethylsilane (TMS) or deuterated chloroform (CDCl₃)], multiplicity (s=singlet, d=doublet, t=triplet, q=quartet, and m=multiplet), integration, and coupling constant (Hz). ¹³C NMR were recorded on a Varian 300, 400, or 500 MHz spectrometer at ambient temperature. Chemical shifts are reported in ppm from CDCl₃ taken as 77.0 ppm. Mass spectra were obtained on Fisons VG Autospec.

Procedure A. General procedure for the BF₃·OEt₂ mediated rearrangement of vinyl acetals: a flame-dried round bottom

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flask containing a magnetic stir bar was charged with the vinyl acetal (0.11 mmol). Under an atmosphere of argon, toluene was added via syringe (0.5 mL) and the mixture cooled to the desired temperature. BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) was added to the solution via syringe and the reaction allowed to stir 12 h. Triethylamine (0.5 mL) was added to quench the reaction and the mixture was poured into saturated aqueous NaHCO₃. The organic layer was separated and the aqueous layer was extracted with Ether (3×10 mL). The combined organic extracts were washed with brine, dried over Na₂SO₄ and concentrated. Diastereoselectivity was determined by ¹H NMR of the unpurified reaction mixture. Analytically pure material was obtained by column chromatography on silica gel.

Procedure B. General procedure for the Me₃Al/BF₃·OEt₂ mediated rearrangement of vinyl acetals: a flame-dried round bottom flask containing a magnetic stir bar was charged with the vinyl acetal (0.11 mmol) and toluene (0.5 mL). Under an atmosphere of argon, Me₃Al was added via syringe (0.22 mL, 2.0 M in PhMe, 0.44 mmol) at -78°C and allowed to stir 2 min at that temperature. $BF_3 \cdot OEt_2$ (16 µL, 0.12 mmol, neat) was then added to the solution via syringe and the reaction allowed to stir 12 h. Triethylamine (0.5 mL) was added to quench the reaction and the mixture was poured into saturated aqueous NaHCO₃. The organic layer was separated and the aqueous layer was extracted with Ether (3×10 mL). The combined organic extracts were washed with brine, dried over Na₂SO₄ and concentrated. Diastereoselectivity was determined by ¹H NMR of the unpurified reaction mixture. Analytically pure material was obtained by column chromatography on silica gel.

3.1.1. [2(2*S**,4*R**,6*S**)]-2-(2-Isopropyl-6-methyl-[1,3] dioxan-4-yl)-1-phenyl-ethanone (*anti*-6d).



According to general procedure A, vinyl acetal (26.2 mg, 0.10 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C produced the product as a 97:3 mixture of isomers (*anti/syn*). Purification by column chromatography on silica gel (1% ethyl acetate/hexane) afforded *anti*-**6d** (24.0 mg, 92%). ¹H NMR (300 MHz, CDCl₃) δ 7.97 (m, 2H), 7.60 (m, 1H), 7.49 (m, 2H), 4.79–4.72 (m, 1H), 4.55 (d, 1H, J=5.4 Hz), 3.95 (m, 1H), 3.53 (dd, 1H, J=6.6, 15.6 Hz), 3.34 (dd, 1H, J=8.1, 15.3 Hz), 1.91 (m, 1H), 1.72 (m, 1H), 1.54 (m, 1H), 1.23 (d, 3H, J=6.3 Hz), 0.90 (d, 3H, J=6.6 Hz); 0.87 (d, 3H, J= 6.6 Hz); ¹³C NMR (75 MHz, CDCl₃ δ 198.3, 137.1, 133.5, 128.9, 128.4, 99.5 689, 68.1, 40.5, 35.6, 33.0, 22.0, 17.4, 17.1; IR (NaCl, neat) 2970, 2931, 1683, 1449, 1373, 1301, 1280, 1213, 1165, 1124, 1076, 1002, 900, 756, 691 cm⁻¹; HRMS (C₁₆H₂₂O₃+H)⁺ calcd 263.1647. Found 263.1655 (FAB+).

3.1.2. [2(2*S**,4*S**,6*S**)]-2-(2-Isopropyl-6-methyl-[1,3] dioxan-4-yl)-1-phenyl-ethanone (*syn*-6d).



According to general procedure B, vinyl acetal (26.2 mg, 0.10 mmol), Me₃Al (0.22 mL, 0.44 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C produced the product as a 10:90 mixture of isomers (anti/syn). Purification by column chromatography on silica gel (1% ethyl acetate/ hexane) afforded syn-6d (21.0 mg, 80%). ¹H NMR (300 MHz, CDCl₃) δ 7.96 (m, 2H), 7.56 (m, 1H), 745 (m, 2H), 4.27–4.18 (m, 2H), 3.77 (m, 1H), 3.38 (dd, 1H, J=6.6, 16.2 Hz), 2.95 (dd, 1H, J=6.0, 16.2 Hz), 1.80-1.69 (m, 2H), 1.31 (m, 1H), 1.22 (d, 3H, J=6.3 Hz), 0.88 (d, 3H, J=6.9 Hz), 0.85 (d, 3H, J=6.9 Hz); ¹³C NMR (J=75 MHz, CDCl₃) δ 198.5, 137.5, 133.4, 128.7, 128.5, 105.6, 73.2, 72.4, 45.0, 39.1, 32.9, 21.8, 17.6, 17.3; IR (NaCl, neat) 2969, 1686, 1449, 1380, 1350, 1193, 1164, 1121, 1108, 1065, 1028, 993, 962, 753, 690 cm^{-1} ; HRMS $(C_{16}H_{22}O_3+H)^+$ calcd 263.1647. Found 263.1646 (FAB+).

3.1.3. [2(2*S**,4*R**,6*S**)]-2-(2-*tert*-Butyl-6-methyl-[1,3] dioxan-4-yl)-1-phenyl-ethanone (*anti*-7aa).



According to general procedure A, vinyl acetal (27.6 mg, 0.10 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C produced the product as a 98:2 mixture of isomers (*anti/syn*). Purifi-cation by column chromatography on silica gel (1% ethyl acetate/hexane) afforded *anti-7aa* (22.5 mg, 82%). ¹H NMR (300 MHz, CDCl₃) δ 7.98 (m, 2H), 7.59 (m, 1H), 7.48 (m, 2H), 4.75 (m, 1H), 4.43 (s, 1H), 3.95 (m, 1H), 3.53 (dd, 1H, *J*=6.6, 15 Hz), 3.30 (dd, 1H, *J*=8.1, 15.3 Hz), 1.90–1.80 (m, 1H), 1.50 (ddd, 1H, *J*=1.5, 2.4, 13.5 Hz), 1.21 (d, 3H, *J*= 6 Hz), 0.85 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 198.2, 137.2, 133.3, 128.8, 128.3, 101.0, 68.9, 69.1, 40.6, 35.7, 35.0, 24.9, 22.1; IR (NaCl, neat) 2976, 2957, 2868, 1684, 1449, 1384, 1361, 1215, 1123, 1040, 995, 755, 691 cm⁻¹; HRMS (C₁₇H₂₄O₃+H)⁺ calcd 277.1804. Found 277.1790 (FAB+).

3.1.4. [2(2*S**,4*S**,6*S**)]-2-(2-*tert*-Butyl-6-methyl-[1,3]dioxan-4-yl)-1-phenyl-ethanone (*syn*-7aa).



According to general procedure B, vinyl acetal (27.6 mg, 0.10 mmol), Me₃Al (0.22 mL, 0.44 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C produced the product as a 4:96 mixture of isomers (*anti/syn*). Purification by column chromatography on silica gel (1% ethyl acetate/hexane) afforded *syn*-**7aa** (25.0 mg, 92%). ¹H NMR (300 MHz, CDCl₃) δ 7.98 (m, 2H), 7.55 (m, 1H), 7.44 (m, 2H), 4.19 (m, 1H), 4.10 (s, 1H), 3.72 (m, 1H), 3.35 (dd, 1H, *J*=6.9, 15.9 Hz), 2.90 (dd, 1H, *J*=6.0, 15.6 Hz), 1.63 (ddd, 1H, *J*= 2.1, 2.1, 12.6 Hz), 1.28 (m, 1H), 1.19 (d, 3H, *J*=6.3 Hz), 0.82 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 199.0, 137.7, 133.3, 128.7, 128.7, 107.1, 73.4, 72.3, 45.1, 39.1, 35.0, 24.9, 21.8; IR (NaCl, neat) 2976, 2958, 2909, 2869, 1685, 1449, 1385, 1361, 1152, 1123, 1048, 1002, 690 cm⁻¹; HRMS (C₁₇H₂₄O₃+H)⁺ calcd 277.1804. Found 277.1799 (FAB+).

3.1.5. [2(2*S* *,4*R* *,6*S* *)]-2-(2-*tert*-Butyl-6-hexanyl-[1,3] dioxan-4-yl)-1-phenyl-ethanone (*anti*-7ba).



According to general procedure A, vinyl acetal (34.6 mg, 0.10 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C produced the product as a 98:2 mixture of isomers (*anti/syn*). Purification by column chromatography on silica gel (1% ethyl acetate/hexane) afforded *anti*-**7ba** (30.0 mg, 87%). ¹H NMR (300 MHz, CDCl₃) δ 7.96 (m, 2H), 7.58 (m, 1H), 7.49 (m, 2H), 4.73 (m, 1H), 4.41 (s, 1H), 3.75 (m, 1H), 3.53 (dd, 1H, *J*=6.6, 15.6 Hz), 3.30 (dd, 1H, *J*=8.1, 15.3 Hz), 1.86 (m, 1H), 1.64–1.29 (m, 9H), 0.95–0.87 (m, 5H), 0.85 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 198.5, 137.2, 133.4, 128.9, 128.4, 101.1, 72.0, 69.0, 40.6, 36.3, 35.1, 34.2, 32.0, 29.4, 25.2, 24.8, 22.9, 14.3; IR (NaCl, neat), 2956, 2929, 2858, 1685, 1449, 1361, 1279, 1215, 1116, 1069, 996, 754, 691 cm⁻¹; HRMS (C₂₂H₃₄O₃+H)⁺ calcd 347.2586. Found 347.2578 (FAB+).

3.1.6. [2(2*S**,4*S**,6*S**)]-2-(2-*tert*-Butyl-6-hexanyl-[1,3] dioxan-4-yl)-1-phenyl-ethanone (*syn*-7ba).



According to general procedure B, vinyl acetal (34.6 mg, 0.10 mmol), Me₃Al (0.22 mL, 0.44 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C produced the product as a 2:98 mixture of isomers (*anti/syn*). Purification by column chromatography on silica gel (1% ethyl acetate/hexane) afforded *syn-***7ba** (31.0 mg, 90%). ¹H NMR (300 MHz, CDCl₃) δ 7.98 (m, 2H), 7.57 (m, 1H), 7.47 (m, 2H), 4.19 (m, 1H), 4.10 (s, 1H), 3.56 (m, 1H), 3.37 (dd, 1H, *J*=6.6, 15.3 Hz), 2.92 (dd, 1H, *J*=5.7, 15.9 Hz), 1.70–1.21 (m, 10H), 0.91–0.83 (m, 5H), 0.83 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 199.0, 137.7, 133.3, 128.7, 128.7, 107.1, 76.2, 73.5, 45.2, 37.5, 36.1, 35.1, 32.0, 29.4, 25.2, 24.9, 22.8, 14.3; IR (NaCl, neat), 2956, 2929, 2858, 1687, 1449, 1361, 1350, 1209, 1114, 1077, 1051, 1001, 752, 690 cm⁻¹; HRMS (C₂₂H₃₄O₃+H)⁺ calcd 347.2586. Found 347.2583 (FAB+).

3.1.7. [2(2*S**,4*R**,6*R**)]-2-(2-*tert*-Butyl-6-cyclohexanyl-[1,3]dioxan-4-yl)-1-phenyl-ethanone (*anti*-7ca).



According to general procedure A, vinyl acetal (34.4 mg, 0.10 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C produced the product as a 98:2 mixture of isomers (*anti/syn*). Purification by column chromatography on silica gel (1% ethyl acetate/hexane) afforded *anti*-**7ca** (33.0 mg, 96%). ¹H NMR (300 MHz, CDCl₃) δ 7.78 (m, 2H), 7.40 (m, 1H), 7.29 (m, 2H), 4.60–4.54 (m, 1H), 4.20 (s, 1H), 3.33 (dd, 1H, *J*=6.6, 15.3 Hz), 3.30 (m, 1H), 3.09 (dd, 1H,

J=7.8, 15.3 Hz), 1.80–0.68 (m, 13H), 0.66 (s, 9H); 13 C NMR (75 MHz, CDCl₃) δ 198.3, 137.2, 133.3, 128.8, 128.3, 101.0, 76.2, 69.0, 43.1, 40.7, 35.3, 31.5, 28.8, 28.6, 26.9, 26.4, 26.2, 24.8; IR (NaCl, neat) 2926, 2853, 1684, 1449, 1214, 1113, 1062, 993, 754, 691 cm⁻¹; HRMS (C₂₂H₃₂O₃+H)⁺ calcd 345.2430. Found 345.2420 (FAB+).

3.1.8. [2(2*S**,4*S**,6*R**)]-2-(2-*tert*-Butyl-6-cyclohexanyl-[1,3]dioxan-4-yl)-1-phenyl-ethanone (*syn*-7ca).



According to general procedure B, vinyl acetal (34.4 mg, 0.10 mmol), Me₃Al (0.22 mL, 0.44 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C produced the product as a 3:97 mixture of isomers (*anti/syn*). Purification by column chromatography on silica gel (1% ethyl acetate/hexane) afforded *syn*-**7ca** (30.0 mg, 87%). ¹H NMR (300 MHz, CDCl₃) δ 7.81 (m, 2H), 7.41 (m, 1H), 7.29 (m, 2H), 3.99 (m, 1H), 3.91 (s, 1H), 3.20 (dd, 1H, *J*=6.9, 15.3 Hz), 3.13 (m, 1H), 2.76 (dd, 1H, *J*=5.7, 15.3 Hz), 1.78–0.70 (m, 13H), 0.65 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 199.1, 138.0, 137.7, 133.3, 128.6, 106.9, 80.4, 73.6, 45.3, 42.9, 35.2, 34.6, 28.8, 28.6, 26.9, 26.4, 26.2, 24.9; IR (NaCl, neat) 2926, 2853, 1687, 1449, 1210, 1118, 1001, 753, 690 cm⁻¹; HRMS (C₂₂H₃₂O₃+H)⁺ calcd 345.2430. Found 345.2435 (FAB+).

3.1.9. [2(2*S**,4*R**,6*S**)]-2-(2-*tert*-Butyl-6-*iso*propyl-[1,3] dioxan-4-yl)-1-phenyl-ethanone (*anti*-7da).



According to general procedure A, vinyl acetal (30.4 mg, 0.10 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C produced the product as a 98:2 mixture of isomers (*anti/syn*). Purification by column chromatography on silica gel (1% ethyl acetate/hexane) afforded *anti*-**7da** (29.5 mg, 97%). ¹H NMR (300 MHz, CDCl₃) δ 7.97 (m, 2H), 7.59 (m, 1H), 7.49 (m, 2H), 4.73 (m, 1H), 4.40 (s, 1H), 3.53 (dd, 1H, *J*=6.6, 15.3 Hz), 3.44 (m, 1H), 3.29 (dd, 1H, *J*=7.8, 15.3 Hz), 1.89 (m, 1H), 1.67 (m, 1H), 1.50 (m, 1H), 0.96 (d, 3H, *J*=6.6 Hz), 0.88 (d, 3H, *J*=6.9 Hz), 0.85 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 198.5, 137.3, 133.4, 128.9, 128.4, 101.0, 77.65, 68.9, 40.6, 35.2, 33.3, 31.3, 24.7, 18.3; IR (NaCl, neat) 2958, 1684, 1449, 1361, 1300, 1214, 1114, 1062, 1000, 754, 691 cm⁻¹; HRMS (C₁₉H₂₈O₃+H)⁺ calcd 305.2117. Found 305.2106 (FAB+).

3.1.10. [2(2*S**,4*S**,6*S**)]-2-(2-*tert*-Butyl-6-isopropyl-[1,3]dioxan-4-yl)-1-phenyl-ethanone (*syn*-7da).



According to general procedure B, vinyl acetal (30.4 mg,

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0.10 mmol), Me₃Al (0.22 mL, 0.44 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C produced the product as a 3:97 mixture of isomers (*anti/syn*). Purification by column chromatography on silica gel (1% ethyl acetate/ hexane) afforded *syn*-**7da** (26.0 mg, 87%). ¹H NMR (300 MHz, CDCl₃) δ 8.00 (m, 2H), 7.57 (m, 1H), 7.46 (m, 2H), 4.18 (m, 1H), 4.09 (s, 1H), 3.38 (dd, 1H, *J*=6.9, 15.3 Hz), 3.28 (m, 1H), 2.93 (dd, 1H, *J*=6.0, 15.6 Hz), 1.74–1.61 (m, 2H), 1.27 (m, 1H), 0.95 (d, 3H, *J*=6.9 Hz), 0.90 (d, 3H, *J*=6.3 Hz), 0.85 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 199.1, 137.7, 133.3, 128.7, 106.9, 81.0, 73.5, 45.3, 35.2, 34.4, 33.2, 24.9, 18.3, 18.3; IR (NaCl, neat) 2959, 1687, 1449, 1114, 1052, 1009, 753, 690 cm⁻¹; HRMS (C₁₉H₂₈O₃+H)⁺ calcd 305.2117. Found 305.2105 (FAB+).

3.1.11. [2(2*S**,4*R**,6*R**)]-2-(2-*tert*-Butyl-6-phenyl-[1,3]-dioxan-4-yl)-1-phenyl-ethanone (*anti*-7ea).



According to general procedure A, vinyl acetal (33.8 mg, 0.10 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C produced the product as a 98:2 mixture of isomers (*anti/syn*). Purification by column chromatography on silica gel (1% ethyl acetate/hexane) afforded *anti-7ea* (29.0 mg, 87%). ¹H NMR (300 MHz, CDCl₃) δ 8.02–7.99 (m, 2H), 7.60 (m, 1H), 7.53–7.48 (m, 2H), 7.37–7.27 (m, 5H), 4.93–4.80 (m, 2H), 4.64 (s, 1H), 3.66 (dd, 1H, *J*=6.9, 15.6 Hz), 3.44 (dd, 1H, *J*=7.8, 15.3 Hz), 2.12 (m, 1H), 1.76 (m, 1H), 0.92 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 198.2, 142.6, 137.2, 133.5, 129.0, 128.5, 128.4, 127.6, 125.6, 101.0, 73.5, 69.1, 40.3, 36.2, 35.3, 24.8; IR (NaCl, neat), 2957, 1683, 1449, 1359, 1215, 1120 cm⁻¹; HRMS (C₂₁H₂₆O₃+H)⁺ calcd 339.1960. Found 339.1976 (FAB+).

3.1.12. [2(2*S**,4*S**,6*R**)]-2-(2-*tert*-Butyl-6-cyclohexanyl-[1,3]dioxan-4-yl)-1-phenyl-ethanone (*syn*-7ea).



According to general procedure B, vinyl acetal (33.8 mg, 0.10 mmol), Me₃Al (0.22 mL, 0.44 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C were stirred for 36 h to produce the product as a 2:98 mixture of isomers (anti/syn). Purification by column chromatography on silica gel (1%) ethyl acetate/hexane) afforded syn-7ea (27.0 mg, 81%). ¹H NMR (300 MHz, CDCl₃) δ 8.00-7.97 (m, 2H), 7.57 (m, 1H), 7.50-7.44 (m, 2H), 7.38-7.23 (m, 5H), 4.72 (m, 1H), 4.39 (m, 1H), 4.34 (s, 1H), 3.41 (ddd, 1H, J=1.2, 6.6, 15.6 Hz), 2.97 (ddd, 1H, J=1.2, 6.0, 15.9 Hz), 1.99 (m, 1H), 1.52 (m, 1H), 0.91 (s, 9H); 13 C NMR (75 MHz, CDCl₃) δ 198.7, 142.6, 137.6, 133.4, 128.7, 128.6, 128.5, 127.5, 125.7, 107.5, 77.65, 73.5, 45.0, 39.7, 35.3, 25.0; IR (NaCl, neat), 2958, 1686, 1483, 1449, 1362, 1210, 1180, 1121, 1057, 1041 cm⁻¹; HRMS (C₂₂H₂₆O₃+H)⁺ calcd 339.1960. Found 339.1969 (FAB+).

3.1.13. [2(2*S**,4*R**,6*S**)]-2-(2-*tert*-Butyl-6-phenylethyl-[1,3]dioxan-4-yl)-1-phenyl-ethanone (*anti*-7fa).



According to general procedure A, vinyl acetal (36.6 mg, 0.10 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C produced the product as a 98:2 mixture of isomers (*anti/syn*). Purification by column chromatography on silica gel (1% ethyl acetate/hexane) afforded *anti*-**7fa** (33.0 mg, 91%). ¹H NMR (300 MHz, CDCl₃) δ 7.96–7.92 (m, 2H), 7.60–7.55 (m, 1H), 7.49–7.45 (m, 2H), 7.31–7.26 (m, 2H), 7.21–7.16 (m, 3H), 4.73 (m, 1H), 4.40 (s, 1H), 3.74 (m, 1H), 3.47 (dd, 1H, *J*=6.6, 15.6 Hz), 2.60 (dd, 1H, *J*=7.5, 15.3 Hz), 2.84–2.62 (m, 2H), 1.88 (m, 2H), 1.70 (m, 1H), 1.47 (m, 1H), 0.89 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 198.4, 142.4, 137.2, 133.4, 128.9, 128.7, 128.6, 128.4, 126.0, 101.0, 70.6, 68.9, 40.6, 37.9, 35.1, 34.1, 24.8; IR (NaCl, neat) 2955, 1684, 1449, 1360, 1123 cm⁻¹; HRMS (C₂₄H₃₀O₃+H)⁺ calcd 367.2273. Found 367.2257 (FAB+).

3.1.14. [2(2*S* *,4*S* *,6*S* *)]-2-(2-*tert*-Butyl-6-phenylethyl-[1,3]dioxan-4-yl)-1-phenyl-ethanone (*syn*-7fa).



According to general procedure B, vinyl acetal (36.6 mg, 0.10 mmol), Me₃Al (0.22 mL, 0.44 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C produced the product as a 4:96 mixture of isomers (anti/syn). Purification by column chromatography on silica gel (1% ethyl acetate/hexane) afforded syn-7fa (31.0 mg, 85%). ¹H NMR (300 MHz, CDCl₃) δ 8.00–7.96 (m, 2H), 7.59–7.53 (m, 1H), 7.48– 7.42 (m, 2H), 7.31–7.26 (m, 2H), 7.21–7.16 (m, 3H), 4.216 (m, 1H), 4.10 (s, 1H), 3..56 (m, 1H), 3.36 (dd, 1H, J=6.3, 15.3 Hz), 2.92 (dd, 1H, J=6.0, 15.9 Hz), 2.72 (m, 2H), 1.88 (m, 1H), 1.74 (m, 1H), 1.65 (m, 1H), 1.33 (m, 1H), 0.89 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 198.9, 142.3, 137.6, 133.3, 128.8, 128.7, 128.6, 128.5, 126.0, 107.0, 74.7, 73.3, 45.1, 37.7, 37.4, 35.2, 31.3, 24.9; IR (NaCl, neat) 2955, 2912, 1686, 1449, 1350, 1209, 1120, 1093, 1047, 1001, 752, 690 cm⁻¹; HRMS $(C_{24}H_{30}O_3+H)^+$ calcd 367.2273. Found 367.2266 (FAB+).





According to general procedure A, vinyl acetal (30.4 mg, 0.10 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C produced the product as a 84:16 mixture of isomers (*anti/syn*). Purification by column chromatography on silica gel (1% ethyl acetate/hexane) afforded *anti*-**7fb** (23.6 mg, 78%). ¹H NMR (300 MHz, CDCl₃) δ 7.31–7.16 (m, 5H), 4.60 (m, 1H), 4.30 (s, 1H), 3.65 (m, 1H), 3.01 (dd, 1H, *J*= 8.4, 15.3 Hz), 2.73 (m, 2H), 2.56 (dd, 1H, *J*=6.9, 15.3 Hz),

2.19 (s, 3H), 1.86 (m, 2H), 1.67 (m, 1H), 1.32 (m, 1H), 0.89 (s, 9H); 13 C NMR (75 MHz, CDCl₃) δ 206.8, 142.1, 128.7, 128.5, 125.9, 100.7, 70.3, 68.3, 45.8, 37.8, 35.1, 34.0, 31.2, 30.2, 24.8; IR (NaCl, neat), 2955, 1716, 1359, 1123, 1097 cm⁻¹; HRMS (C₁₉H₂₈O₃+H)⁺ calcd 305.2117. Found 305.2114 (FAB+).

3.1.16. [2(2*S* *,4*S* *,6*S* *)]-2-(2-*tert*-Butyl-6-phenylethyl-[1,3]dioxan-4-yl)-1-methyl-ethanone (*syn*-7fb).



According to general procedure B, vinyl acetal (30.4 mg, 0.10 mmol), Me₃Al (0.22 mL, 0.44 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -55°C produced the product as a 4:96 mixture of isomers (anti/syn). Purification by column chromatography on silica gel (1% ethyl acetate/ hexane) afforded syn-7fb (27.0 mg, 89%). ¹H NMR (300 MHz, CDCl₃) δ 7.31-7.26 (m, 2H), 7.20-7.16 (m, 3H), 4.09 (s, 1H), 3.98 (m, 1H), 3.52 (m, 1H), 2.82-2.63 (m, 3H), 2.44 (dd, 1H, J=4.5, 14.7 Hz), 2.19 (s, 3H), 1.86 (m, 1H), 1.70 (m, 1H), 1.49 (m, 1H), 1.21 (m, 1H), 0.92 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 207.7, 142.2, 128.7, 128.5, 126.0, 106.9, 74.7, 73.0, 49.9, 37.6, 37.1, 35.2, 31.4, 31.3, 25.0; IR (NaCl, neat), 2956, 1717, 1484, 1455, 1361, 1217, 1155, 1119, 1097, 1079, 1044, 1005 cm⁻¹; HRMS $(C_{19}H_{28}O_3+H)^+$ calcd 305.2117. Found 305.2121 (FAB+).

3.1.17. [2(2*S**,4*R**,6*S**)]-2-(2-*tert*-Butyl-6-phenylethyl-[1,3]dioxan-4-yl)-1-ethyl-ethanone (*anti*-7fc).



According to general procedure A, vinyl acetal (31.8 mg, 0.10 mmol) and BF₃·OEt₂ (16 μL, 0.12 mmol, neat) at -78° C produced the product as a 96:4 mixture of isomers (*anti/syn*). Purification by column chromatography on silica gel (1% ethyl acetate/hexane) afforded *anti*-**7fc** (29.0 mg, 91%). ¹H NMR (300 MHz, CDCl₃) δ 7.31–7.26 (m, 2H), 7.21–7.16 (m, 3H), 4.60 (m, 1H), 4.31 (s, 1H), 3.65 (m, 1H), 2.99 (dd, 1H, *J*=8.1, 14.1 Hz), 2.73 (m, 2H), 2.58–2.04 (m, 3H), 1.85 (m, 2H), 1.68 (m, 1H), 1.33 (m, 1H), 1.04 (t, *J*=7.2 Hz), 0.89 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 209.1, 142.1, 128.6, 128.5, 125.9, 100.6, 70.4, 68.5, 44.6, 37.6, 36.4, 35.2, 34.2, 31.3, 24.9, 7.9; IR (NaCl, neat), 2976, 1715, 1455, 1360, 1123, 1074, 1040, 996 cm⁻¹; HRMS (C₂₀H₃₀O₃+H)⁺ calcd 319.2273. Found 319.2266 (FAB+).

3.1.18. [2(2*S* *,4*S* *,6*S* *)]-2-(2-*tert*-Butyl-6-phenylethyl-[1,3]dioxan-4-yl)-1-ethyl-ethanone (*syn*-7fc).



According to general procedure B, vinyl acetal (31.8 mg, 0.10 mmol), Me₃Al (0.22 mL, 0.44 mmol) and $BF_3 \cdot OEt_2$

(16 μL, 0.12 mmol, neat) at -55° C produced the product as a 2:98 mixture of isomers (*anti/syn*). Purification by column chromatography on silica gel (1% ethyl acetate/hexane) afforded *syn*-**7fc** (28.0 mg, 89%). ¹H NMR (300 MHz, CDCl₃) δ 7.30–7.25 (m, 2H), 7.20–7.16 (m, 3H), 4.08 (s, 1H), 3.98 (m, 1H), 3.51 (m, 1H), 2.82–2.63 (m, 3H), 2.57– 2.37 (m, 3H), 1.86 (m, 1H), 1.69 (m, 1H), 1.50 (m, 1H), 1.23 (m, 1H), 1.04 (t, 3H, *J*=7.5 Hz), 0.91 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 210.2, 142.2, 128.7, 128.5, 125.9, 106.9, 74.7, 73.1, 48.7, 37.6, 37.6, 37.2, 35.1, 31.3, 25.0, 7.7; IR (NaCl, neat), 2976, 2955, 2867, 1716, 1483, 1455, 1377, 1361, 1342, 1133, 1095, 1080, 1047, 1004, 748, 700 cm⁻¹; HRMS (C₂₀H₃₀O₃+H)⁺ calcd 319.2273. Found 319.2277 (FAB+).

3.1.19. [2(2*S**,4*S**,5*S**,6*S**)]-2-(2-*tert*-Butyl-5-methyl-6-methyl-[1,3]dioxan-4-yl)-1-phenyl-ethanone (16).



According to general procedure B, vinyl acetal (29.0 mg, 0.10 mmol), Me₃Al (0.22 mL, 0.44 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C produced the product as a 4:96 mixture of isomers (*anti/syn*). Purification by column chromatography on silica gel (1% ethyl acetate/hexane) afforded **16** (23.5 mg, 81%). ¹H NMR (300 MHz, CDCl₃) δ 7.99 (m, 2H), 7.53 (m, 1H), 7.45 (m, 2H), 4.08 (s, 1H), 3.83 (m, 1H), 3.31 (m, 2H), 2.94 (dd, 1H, *J*=3.0, 14.1 Hz), 1.36 (m, 1H), 1.22 (d, 3H, *J*=6.0, Hz), 0.84 (d, 3H, *J*=6.6 Hz), 0.74 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 200.0, 138.1, 133.1, 128.9, 128.5, 106.7, 79.4, 78.0, 42.7, 41.3, 34.9, 24.8, 19.6, 12.6; IR (NaCl, neat), 2976, 1686, 1449, 1362, 1260, 1145, 1124, 1093, 1048 cm⁻¹; HRMS (C₁₈H₂₆O₃+H)⁺ calcd 291.1960. Found 291.1951 (FAB+).

3.1.20. [2(2*S* *,4*R* *,5*S* *,6*S* *)]-2-(2-*tert*-Butyl-5-methyl-6-methyl-[1,3]dioxan-4-yl)-1-phenyl-ethanone (17).



According to general procedure A, vinyl acetal (29.0 mg, 0.10 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C produced the product as a 98:2 mixture of isomers (*anti/syn*). Purification by column chromatography on silica gel (1% ethyl acetate/hexane) afforded **17** (36.0 mg, 89%). ¹H NMR (300 MHz, CDCl₃) δ 7.92 (m, 2H), 7.57 (m, 1H), 7.47 (m, 2H), 4.63 (m, 1H), 4.44 (s, 1H), 3.60 (m, 1H), 3.46 (dd, 1H, *J*=9.9, 15.0 Hz), 3.03 (dd, 1H, *J*=4.8, 15.0 Hz), 1.94 (m, 1H), 1.19 (d, 3H, *J*=6.0 Hz), 0.79 (d, 3H, *J*=7.2 Hz), 0.76 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 199.1, 137.5, 133.2, 128.9, 128.4, 100.4, 73.5, 73.4, 39.0, 36.4, 34.8, 24.7, 19.6, 13.0; IR (NaCl, neat), 2975, 2959, 1684, 1448, 1359, 1287, 1216, 1136, 1123, 1090, 1045 cm⁻¹; HRMS (C₁₈H₂₆O₃+H)⁺ calcd 291.1960. Found 291.1956 (FAB+).

3.1.21. [2(2*S**,4*R**,5*R**,6*S**)]-2-(2-*tert*-Butyl-5-methyl-6-methyl-[1,3]dioxan-4-yl)-1-phenyl-ethanone (19).

According to general procedure A, vinyl acetal (29.0 mg, 0.10 mmol) and BF₃·OEt₂ (16 μ L, 0.12 mmol, neat) at -78° C produced the product as a 98:2 mixture of isomers (*anti/syn*). Purification by column chromatography on silica gel (1% ethyl acetate/hexane) afforded **19** (26.0 mg, 90%). ¹H NMR (300 MHz, CDCl₃) δ 7.97 (m, 2H), 7.59 (m, 1H), 7.49 (m, 2H), 4.44 (m, 1H), 4.44 (s, 1H), 4.09 (m, 1H), 3.56 (dd, 1H, *J*=6.9, 15.6 Hz), 3.33 (dd, 1H, *J*=7.5, 15.6 Hz), 1.40 (m, 1H), 1.15 (d, 3H, *J*=2.1 Hz), 1.13 (d, 3H, *J*=1.5 Hz), 0.85 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 198.5, 137.3, 133.4, 128.9, 128.4, 101.3, 75.9, 69.9, 40.2, 35.4, 35.0, 24.7, 18.9, 12.9; IR (NaCl, neat), 2977, 1684, 1449, 1328, 1283, 1216, 1083 cm⁻¹; HRMS (C₁₈H₂₆O₃+H)⁺ calcd 291.1960. Found 291.1955 (FAB+).

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References

- 1. Rychnovsky, S. D. Chem. Rev. 1995, 95, 2021.
- For a review, see: (a) Schneider, C. Angew. Chem., Int. Ed. 1998, 37, 1375. (b) Sinz, C. J.; Rychnovsky, S. D. Top. Curr. Chem. 2001, 216, 51. (c) Mori, Y.; Asai, M.; Kawade, J.-I.; Furukawa, H. Tetrahedron 1995, 51, 5315. (d) Nicolaou, K. C.; Daines, R. A.; Uenishi, J.; Li, W. S.; Papahatjis, D. P.; Chakraborty, T. K. J. Am. Chem. Soc. 1988, 110, 4672. (e) Poss, C. S.; Schreiber, S. L. Acc. Chem. Res. 1994, 27, 9.

(f) Rychnovsky, S. D.; Khire, U. R.; Yang, G. J. Am. Chem. Soc. 1997, 119, 2058. (g) Mori, Y.; Kuhara, M.; Takeuchi, A.; Suzuki, M. Tetrahedron Lett. 1988, 5419. (h) Mori, Y.; Suzuki, M. Tetrahedron Lett. 1989, 30, 4383. (i) Zacuto, M. J.; Leighton, J. L. J. Am. Chem. Soc. 2000, 122, 8587. (j) Zacuto, M. J.; O'Malley, S. J.; Leighton, J. L. J. Am. Chem. Soc. 2002, 124, 7890.

- Zhang, Y.; Reynolds, N. T.; Manju, K.; Rovis, T. J. Am. Chem. Soc. 2002, 124, 9720.
- (a) Crich, D.; Jiao, X.-Y.; Brunko, M. *Tetrahedron* 1997, *53*, 7127. (b) Kurihara, M.; Miyata, N. *Chem. Lett.* 1995, 263. (c) Tsunoda, T.; Suzuki, M.; Noyori, R. *Tetrahedron Lett.* 1980, *21*, 1357.
- (a) Dahanukar, V. H.; Rychnovsky, S. D. J. Org. Chem. 1996, 61, 8317. (b) Kopecky, D. J.; Rychnovsky, S. D. J. Org. Chem. 2000, 65, 191.
- (a) Petasis, N. A.; Bzowej, E. I. J. Am. Chem. Soc. 1990, 112, 6392. (b) Petasis, N. A.; Lu, S.-P.; Bzowej, E. I.; Fu, D.-K.; Staszewski, J. P.; Akritopoulou-Zanze, I.; Patane, M. A.; Hu, Y.-H. Pure Appl. Chem. 1996, 68, 667. (c) Payack, J. F.; Hughes, D. L.; Cai, D.; Cottrell, I. F.; Verhoeven, T. R. Org. Synth. 2002, 79, 19.
- (a) Hogen-Esch, T. E.; Smid, J. J. Am. Chem. Soc. 1965, 87, 669.
 (b) Hogen-Esch, T. E.; Smid, J. J. Am. Chem. Soc. 1965, 88, 307.
 (c) Hogen-Esch, T. E.; Smid, J. J. Am. Chem. Soc. 1965, 88, 318.
 (d) Buncel, E.; Menon, B. J. Org. Chem. 1979, 44, 317.
- See, for example: (a) Davies-Coleman, M. T.; Garson, M. J. Nat. Prod. Rep. 1998, 15, 477. (b) Rohr, J. Angew. Chem., Int. Ed. 2000, 39, 2847. (c) Omura, S. Macrolide Antibiotics. Academic: New York, 1984.
- See, for example: (a) Marshall, J. A.; Schaaf, G. M. J. Org. Chem. 2001, 66, 7825. (b) Evans, D. A.; Carter, P. A.; Carreira, E. M.; Prunet, J. A.; Charette, A. B.; Lautens, M. Angew. Chem., Int. Ed. 1998, 37, 2354. (c) Scheidt, K. A.; Tasaka, A.; Bannister, T. D.; Wendt, M. D.; Roush, W. R. Angew. Chem., Int. Ed. 1999, 38, 1652. (d) Guo, J. S.; Duffy, K. J.; Stevens, K. L.; Dalko, P. I.; Roth, R. M.; Hayward, M. M.; Kishi, Y. Angew. Chem., Int. Ed. 1998, 37, 187. (e) Panek, J. S.; Jain, N. F. J. Org. Chem. 1998, 63, 4572. (f) BouzBouz, S.; Popkin, M. E.; Cossy, J. Org. Lett. 2000, 2, 3449.